

AP BIO UNIT 2 CHEAT SHEET

AP Bio Unit 2 Cheat Sheet: As students dive into the world of Advanced Placement Biology, mastering the material covered in Unit 2 is crucial. This unit focuses on the chemistry of life, providing a foundation for understanding biological processes. With a plethora of information to digest, having a cheat sheet can be an invaluable resource for quick reference and review. In this article, we will break down the key concepts, important molecules, and essential processes that you need to know for Unit 2, ensuring that you are well-prepared for your AP Biology exam.

Overview of AP Biology Unit 2

AP Biology Unit 2 centers around the chemistry that underpins biological systems. Understanding how atoms and molecules interact is fundamental to grasping larger biological concepts. Here are some of the core topics covered in this unit:

- **Water and its properties**
- **Macromolecules:** Carbohydrates, proteins, lipids, and nucleic acids
- **Enzymes and their functions**
- **The role of pH and temperature in biological processes**

Knowing these topics will not only help you in the exam but also in understanding the more complex systems in later units.

Key Concepts in Unit 2

Water: The Essence of Life

Water is a polar molecule with unique properties that make it essential for life. Understanding these properties is critical for any biology student.

- **Polarity:** Water molecules have a partial positive charge on one side and a partial negative charge on the other, leading to hydrogen bonding.
- **Cohesion:** The attraction between water molecules allows for surface tension, which is vital for many biological processes.
- **Adhesion:** Water's ability to stick to other substances is important for processes like capillary action in plants.
- **High Specific Heat:** Water can absorb a lot of heat before increasing in temperature, which helps regulate climate and maintain stable environments for organisms.

MACROMOLECULES: BUILDING BLOCKS OF LIFE

AP BIOLOGY UNIT 2 DIVES DEEP INTO THE FOUR MAJOR MACROMOLECULES THAT ARE ESSENTIAL FOR LIFE. EACH MACROMOLECULE HAS UNIQUE STRUCTURES AND FUNCTIONS.

1. **CARBOHYDRATES:** COMPOSED OF CARBON, HYDROGEN, AND OXYGEN, CARBOHYDRATES PROVIDE ENERGY AND STRUCTURAL SUPPORT. THEY CAN BE CLASSIFIED INTO MONOSACCHARIDES (SIMPLE SUGARS), DISACCHARIDES (TWO SUGARS), AND POLYSACCHARIDES (MANY SUGARS).
2. **PROTEINS:** MADE UP OF AMINO ACIDS, PROTEINS PERFORM A VAST ARRAY OF FUNCTIONS INCLUDING CATALYZING REACTIONS AS ENZYMES, PROVIDING STRUCTURAL SUPPORT, AND FACILITATING TRANSPORT.
3. **LIPIDS:** THESE HYDROPHOBIC MOLECULES INCLUDE FATS, OILS, AND PHOSPHOLIPIDS, PLAYING CRITICAL ROLES IN ENERGY STORAGE, CELL MEMBRANE STRUCTURE, AND SIGNALING.
4. **NUCLEIC ACIDS:** DNA AND RNA ARE THE MOLECULES OF HEREDITY, RESPONSIBLE FOR STORING AND TRANSMITTING GENETIC INFORMATION.

ENZYMES: BIOLOGICAL CATALYSTS

ENZYMES ARE PROTEINS THAT SPEED UP CHEMICAL REACTIONS WITHOUT BEING CONSUMED IN THE PROCESS. UNDERSTANDING THEIR FUNCTION IS VITAL FOR AP BIOLOGY.

- **ACTIVE SITE:** THE REGION ON AN ENZYME WHERE SUBSTRATE MOLECULES BIND AND UNDERGO A CHEMICAL REACTION.
- **SUBSTRATE:** THE REACTANT MOLECULE UPON WHICH AN ENZYME ACTS.
- **ENZYME-SUBSTRATE COMPLEX:** THE TEMPORARY COMPLEX FORMED WHEN AN ENZYME BINDS TO ITS SUBSTRATE.
- **FACTORS AFFECTING ENZYME ACTIVITY:** TEMPERATURE, pH, AND SUBSTRATE CONCENTRATION CAN ALL INFLUENCE THE RATE OF ENZYME-CATALYZED REACTIONS.

UNDERSTANDING pH AND TEMPERATURE

THE FUNCTION OF BIOLOGICAL MOLECULES IS OFTEN INFLUENCED BY ENVIRONMENTAL CONDITIONS SUCH AS pH AND TEMPERATURE. HERE, WE'LL DISCUSS HOW THESE FACTORS IMPACT BIOLOGICAL PROCESSES.

THE ROLE OF pH